

General Approach for the Synthesis of Indole Alkaloids via the Asymmetric Pictet–Spengler Reaction: First Enantiospecific Total Synthesis of (–)-Corynantheidine as Well as the Enantiospecific Total Synthesis of (–)-Corynantheidol, (–)-Geissoschizol, and (+)-Geissoschizine

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Received May 15, 2000

Corynantheidine **1** was first isolated in 1944 by Janot et al. from the African plant *Pseudocinchona africana*,¹ and the structure was determined by the same group.² A number of partial and total syntheses of **1** have been reported, all of which resulted in racemates.³ The related corynantheidol **2** was obtained from *Mitragyna parvifolia* (Roxb.) Korth. (Rubiaceae).^{4a} Although several total syntheses of **2** have been realized,⁵ only one approach was enantioselective (up to 86% ee); Meyers and co-workers completed this route in 1991 in excellent overall yield (16.4%).⁶ Geissoschizol **3** has been isolated from *Hunteria zeylanica* var. *africana*,⁷ and many elegant syntheses have been reported,^{5b,8} at least one of which was enantioselective.^{8a} Geissoschizine **4**, historically one of the most important intermediates in the biosynthesis of monoterpene indole alkaloids,^{10a} has been obtained from a number of plants.¹¹ Because of the biosynthetic importance of **4**, its structural complexity, and the scarce availability from natural sources, there have been many important total syntheses of this natural product.^{9,10} Among these, those of Winterfeldt, Overman, and Martin were enantioselective.^{8a,10}

In this contribution, the first enantiospecific total synthesis of (–)-corynantheidine (**1**) as well as an efficient enantiospecific total synthesis of (–)-corynantheidol (**2**), (–)-geissoschizol (**3**), and (+)-geissoschizine (**4**) from a common intermediate are described (Figure 1). The stereochemical integrity of these natural products was guaranteed via the *trans* transfer of asymmetry via a new extension of the asymmetric¹² Pictet–Spengler reaction.

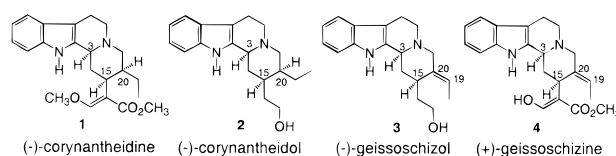
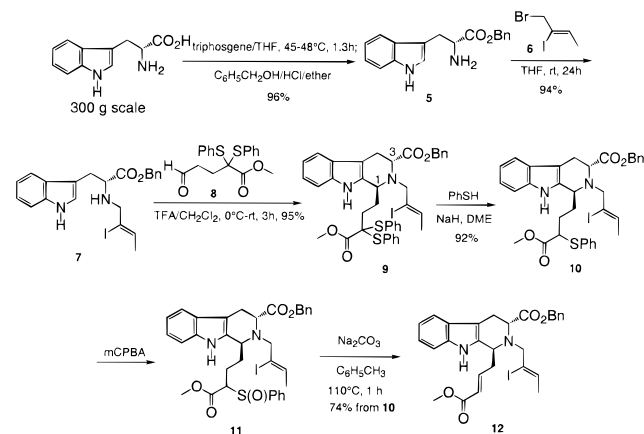


Figure 1. Structure of (–)-corynantheidine, (–)-corynantheidol, (–)-geissoschizol, and (+)-geissoschizine.

The synthesis of the common intermediate is outlined in Scheme 1. The benzyl ester of D-tryptophan (**5**) was prepared on 300 g scale in 96% yield according to a modified literature procedure.¹³ Monoalkylation of the N₆-amino moiety with allylic

Scheme 1. Preparation of the Common Intermediate **12**



bromide **6**, a building block prepared earlier by Ensley^{14a} and employed by Bosch,^{14b} Rawal,^{8b,15} as well as Kuehne,¹⁶ was achieved in excellent yield, employing **5** and **6** at high concentration in the presence of a slight excess of the benzyl ester **5**. The stereospecific, enantiospecific construction of the chiral center at C-1¹² in **9** was achieved by employing a modification of the asymmetric Pictet–Spengler reaction.¹² When a solution of aldehyde **8**¹⁷ and 1 equiv of benzyl ester **7** was stirred in methylene chloride in the presence of TFA, this provided the tetrahydro- β -carboline **9** in 95% yield with complete *trans*-transfer of chirality¹² from C-3 to C-1. No *cis*-isomer was detected under these conditions. In Overman's elegant synthesis of (+)-geissoschizine,^{10a} a similar reaction was attempted wherein the N₆-nitrogen atom in **7** was devoid of the alkyl substituent. In that case, a moderate (40%) yield was reported with a *trans*- to *cis*- ratio of approximately 1:4. The results (see **9**) described herein provide further evidence of the strong directing and accelerating effect of the large alkyl group¹² on the N₆-nitrogen atom on the stereoselectivity of the asymmetric Pictet–Spengler reaction.¹² This suggests that a bulky substituent on the N₆-nitrogen atom is the only requirement necessary to achieve 100% diastereoselectivity in the Pictet–Spengler reaction of carbonyl compounds with tryptophan alkyl esters. With tetrahydro- β -carboline **9** in hand, the desired α,β -unsaturated ester **12** was readily prepared in good yield via a series of standard transformations including removal of one equivalent of thiophenol from **9** followed by an oxidation (see **10** \rightarrow **11**), sulfoxide elimination sequence (see Scheme 1).¹⁷

(13) Wilchek, M.; Patchornik, A. *J. Org. Chem.* **1963**, *28*, 1874.

(14) (a) Ensley, H. E.; Buescher, R. R.; Lee, K. *J. Org. Chem.* **1982**, *47*, 404. (b) Bonjoch, J.; Sole, D.; Garcia-Rubio, S.; Bosch, J. *J. Am. Chem. Soc.* **1997**, *119*, 7230.

(15) Rawal, V. H.; Michoud, C.; Monsted, R. *J. Am. Chem. Soc.*, **1993**, *115*, 3030.

(16) Kuehne, M. E.; Wang, T.; Seraphin, D. *J. Org. Chem.*, **1996**, *61*, 7873.

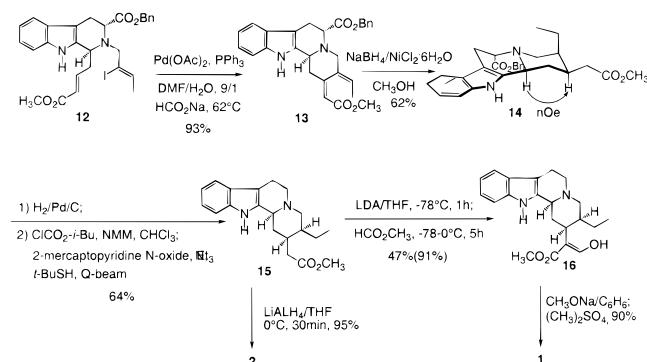
(17) Massiot, G.; Mulamba, T. *J. Chem. Soc., Chem. Commun.* **1983**, 1147.

- (1) Janot, M.-M.; Goutarel, R. *C. R. Acad. Sci.* **1944**, *218*, 852.
 (2) Janot, M.-M.; Goutarel, R.; Le Hir, A.; Tsatsas, G.; Prelog, V. *Helv. Chim. Acta.* **1955**, *38*, 1073.
 (3) (a) Weisbach, J. A.; Kirkpatrick, J. L.; Williams, K. R.; Anderson, E. L.; Yim, N. C.; Douglas, B. *Tetrahedron Lett.* **1965**, 3457. (b) Wenkert, E.; Dave, K. G.; Lewis, R. G.; Sprague, P. W. *J. Am. Chem. Soc.* **1967**, *89*, 6741. (c) Szántay, C.; Bárczai-Beke, M. *Chem. Ber.* **1969**, *102*, 3693. (d) Brown, R. T.; Chapple, C. L.; Charalambides, A. A. *J. Chem. Soc., Chem. Commun.* **1974**, 756. (e) Van Tamelen, E. E.; Dorschel, C. *Bioorg. Chem.* **1976**, *5*, 203. (f) Sakai, S.; Shinma, N. *Chem. Pharm. Bull.* **1978**, *26*, 2596. (g) Lounasmaa, M.; Jokela, R.; Laine, C.; Hanhinen, P. *Tetrahedron Lett.* **1995**, *36*, 8687.
 (4) (a) Shellard, E. J.; Houghton, P. J. *Planta Med.* **1973**, *24*, 13. (b) Vamvacas, C.; Phillipsborn, W. V.; Schlittler, E.; Schmid, H.; Karrer, P. *Helv. Chim. Acta.* **1957**, *40*, 1793.
 (5) (a) Imanishi, T.; Inoue, M.; Wada, Y.; Hanaoka, M. *Chem. Pharm. Bull.* **1982**, *30*, 1925. (b) Lounasmaa, M.; Jokela, R.; Tirkkonen, B.; Miettinen, J.; Halonen, M. *Heterocycles* **1992**, *34*, 321.
 (6) Beard, R. L.; Meyers, A. I. *J. Org. Chem.* **1991**, *56*, 2091.
 (7) Lavaud, C.; Massiot, G.; Vercauteren, J.; Le Men-Olivier, L. *Phytochemistry* **1982**, *21*, 445.
 (8) (a) Bohlmann, C.; Bohlmann, R.; Rivera, E. G.; Vogel, C.; Manandhar, M. D.; Winterfeldt, E. *Liebigs Ann. Chem.* **1985**, 1752. (b) Birman, V. B.; Rawal, V. H. *Tetrahedron Lett.* **1998**, *39*, 7219.
 (9) Takayama, H.; Watanabe, F.; Kitajima, M.; Aimi, N. *Tetrahedron Lett.* **1997**, *38*, 5307.
 (10) (a) Overman, L. E.; Robichaud, A. J. *J. Am. Chem. Soc.* **1989**, *111*, 300. (b) Martin, S. F.; Chen, K. X.; Eary, C. T. *Org. Lett.* **1999**, *1*, 79.
 (11) (a) Puisieux, F.; Goutarel, R.; Janot, M. M.; LeHir, A. C. R. *Seances Acad. Sci. Ser. 2* **1959**, 249, 1369. (b) Rapoport, H.; Windgasson, R. J., Jr.; Hughes, N. A.; Onak, T. P. *J. Am. Chem. Soc.* **1960**, *82*, 4404. (c) Janot, M.-M.; *Tetrahedron* **1961**, *14*, 113. (d) Mehri, H.; Sciamama, M.; Pliat, T.; Sevenet, T.; Pusset, J. *Ann. Pharm. Fr.* **1984**, *42*, 145.
 (12) Cox, E. D.; Hamaker, L. K.; Li, J.; Yu, P.; Czerwinski, K. M.; Deng, L.; Bennett, D. W.; Cook, J. M. *J. Org. Chem.* **1997**, *62*, 44.

The synthesis of key intermediate **12** required six steps from D-(+)-tryptophan with an overall yield of 58%.

With the α,β -unsaturated ester **12** in hand, attention turned to the construction of the all *cis*-D-ring. Intramolecular Heck coupling of intermediate **12**, analogous to the coupling process in other systems^{8b} provided $\alpha,\beta,\gamma,\delta$ -unsaturated ester **13** in near quantitative yield. After a number of unsuccessful attempts, this ester **13** was successfully reduced with NaBH₄ in the presence of a catalytic amount of NiCl₂·6H₂O¹⁸ to provide **14** in 62% yield, with the partially reduced β,γ -unsaturated ester as the minor byproduct. The all *cis*-relationship among the hydrogen atoms at C-3, C-15, and C-20 was confirmed by nOe studies. In addition, removal of the carboxybenzyl group via catalytic debenzoylation, followed by Barton–Crich decarboxylation¹⁹ provided previously known ester **15** whose spectroscopic properties were in agreement in all respects to those reported earlier.^{3c,e} Reduction of ester **15** with LiAlH₄ at 0 °C afforded (–)-corynantheidol **2** ($[\alpha]_D -102^\circ$; lit. -99° ,^{4b} -93°) in 95% yield, the spectroscopic properties of which were identical in all respects to those reported in the literature.⁶ The synthesis of **2** required 11 steps in 20% overall yield. Treatment of ester **15** with 3 equiv of LDA at -78°C for 1 h followed by addition of excess methyl formate at -78°C after which the mixture was allowed to warm to 0 °C over a period of 5 h furnished a 47% yield of the desired enol **16**. This ester was isolated as a mixture of the enol and the formyl tautomers, moreover the recovered starting material (48%) could be recycled if desired. The 91% yield depicted in Scheme 2 for this process

Scheme 2. Synthesis of Corynantheidol and Corynantheidine

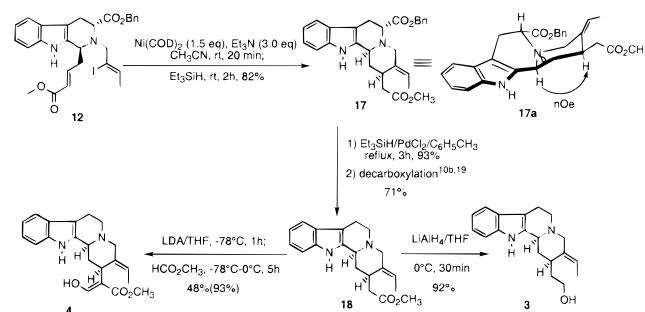


is based on recovered starting **15**. Enol ester **16** was directly methylated using 1 equiv of sodium methoxide and 1 equiv of dimethyl sulfate to provide (–)-corynantheidine **1**. The optical rotation ($[\alpha]_D -166^\circ$, lit. -171° ,² -155° ^{3c}) and spectroscopic properties of synthetic (–)-**1** agree in all respects to those of natural (–)-corynantheidine.³ The route employed 12 steps with an overall yield of 18%. To the best of our knowledge, this is the first enantiospecific total synthesis of (–)-**1**.

To construct the molecular framework of geissoschizol/geissoschizine from **12**, a stereoselective Michael reaction was proposed. Unfortunately, this seemingly simple transformation turned out to be difficult. Because the construction of the D-ring in the synthesis of racemic geissoschizol/geissoschizine had employed transition-metal chemistry, (see the work of Rawal^{8b,15}

and Takayama⁹), it was felt a nickel-mediated process related to the work of Takayama⁹ might provide a solution. Indeed, when vinyl iodide **12** was treated with 1.5 equiv of Ni(COD)₂ and 3.0 equiv of triethylamine in acetonitrile and this was followed by the addition of 2 equiv of Et₃SiH,^{9,14b,20} an 82% yield of the desired *Corynanthe* skeleton **17** was isolated. The important *cis*-relationship between the hydrogen atoms at C-3 and C-15 was confirmed by nOe studies, while the E-configuration of the double bond in **17** was felt to be maintained^{9,14b–16} and later demonstrated (see below). Removal of the benzyl group mediated by Et₃SiH in the presence of PdCl₂²¹ afforded the corresponding carboxylic acid which was converted into the known optically active ester **18** in 71% yield by a Barton–Crich decarboxylation.¹⁹ This intermediate **18** was converted into the two natural products (–)-geissoschizol **3** and (+)-geissoschizine **4** (see Scheme 3). Treatment of **18** with

Scheme 3. Synthesis of Geissoschizine



LiAlH₄ at 0 °C for 30 min afforded (–)-**3** whose optical rotation ($[\alpha]_D -68^\circ$; lit. -70° ,^{11a} -54° ^{8a}) and spectroscopic properties were in full agreement with the reported values.^{5b,8b} This synthesis was completed in 10 steps with an overall yield of 29%. Formylation of **18** under similar conditions to those employed for **15** gave (+)-geissoschizine **4** in 48% yield (93% based on recovered starting material¹⁰). The spectroscopic properties and optical rotation ($[\alpha]_D +113^\circ$; lit. $+114^\circ$,^{11a} $+114^\circ$,^{8a} $+113^\circ$,^{10a} $+109^\circ$ ^{10b}) of synthetic (+)-**4** agree in all respects with those reported for the natural product.^{9–11} This synthesis required 10 steps from D-tryptophan and was completed in an overall yield of 29%. To the best of our knowledge, this is most efficient total synthesis of (+)-geissoschizine reported, to date.

In summary, the first enantiospecific total synthesis of (–)-corynantheidine **1** was achieved (18% overall yield) as well as the enantiospecific synthesis of (–)-corynantheidol, (–)-geissoschizol, and (+)-geissoschizine. A facile entry into key intermediate **12** (the branching point of corynantheidine and geissoschizine) from D-tryptophan was developed in six steps with an overall yield of 58%. This approach extends the scope of the asymmetric Pictet–Spengler reaction while providing a simple route to the enantiomer (from L-tryptophan) of these alkaloids for biological screening. The transition metal mediated formation of *E*-ethylidene moieties for a number of indole alkaloids in this series.

Supporting Information Available: Experimental procedures for **1**, **2**, **3**, **4**, **5**, **7**, **9**, **12**, **13**, **14**, **15**, **17**, and **18** (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(20) Sole, D.; Cancho, Y.; Llebaria, A.; Moreto, J. M.; Delgado, A. *J. Org. Chem.* **1996**, *61*, 5895.

(21) Birkofer, L.; Bierwirth, E.; Ritter, A. *Chem. Ber.* **1961**, *94*, 821.

(18) Jacobi, P. A.; Craig, T. A.; Walker, D. G.; Arrick, B. A.; Frechette, R. F. *J. Am. Chem. Soc.* **1984**, *106*, 5585.

(19) Barton, D. H. R.; Crich, D.; Motherwell, W. B. *Tetrahedron* **1985**, *41*, 3901.